

M1.(a) C_6H_{14}

1

(b) **A**

1

(c) **B**

1

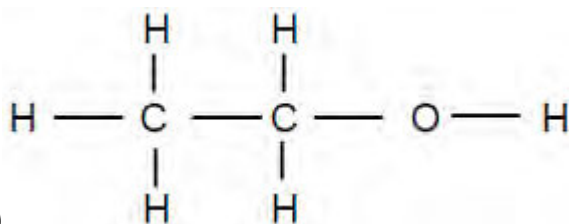
(d) **C**

1

(e) Propanol

1

[5]



M2.(a) (i)

allow other arrangements provided connectivity is correct
allow — OH

1

(ii) oxygen

accept O₂
allow O

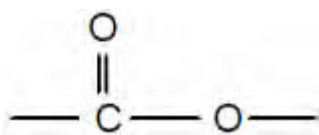
1

oxidation

allow oxidisation / oxidising / oxidised
allow redox

1

(b) (i) ring around



1

(ii) ester(s)

*do **not** allow ether(s)*

1

(iii) propanol

propanol accept propan-1-ol
allow propyl alcohol

1

[6]

- M3.** (a) kills bacteria / sterilises (water)
allow kills microorganisms / microbes / germs
allow 'makes (water) safe (to drink)' or disinfectant
ignore cleans water or removes impurities / bacteria 1
- (b) goes colourless / decolourised (from red / red-brown / brown / yellow / orange)
allow colour disappears
ignore 'goes clear' or discoloured
do not accept incorrect initial colour
do not accept precipitate 1
- (c) (i) Br_2 and 2Cl^-
allow multiples / fractions if whole equation balanced 1
- (ii) changes to red / red-brown / brown / yellow / orange
do not accept effervescence / fizzing / precipitate / gas given off
ignore vapour / temperature changes / ignore initial colour 1
- (d) (i) 7 outer electrons **or**
 same number of outer electrons
allow last / final shell for outer
allow energy level / orbit / ring for shell
allow 'need to gain 1 e⁻ to have a full outer shell'
ignore 'similar number of outer electrons' 1
- (ii) bromine / it (atom) is bigger **or**
must be a comparison

outer electrons (level / shell) further from nucleus **or** more shells

*do **not** accept more outer shells*

ignore more electrons

forces / attractions are weaker **or** more shielding **or** attracts less

*do **not** accept magnetic / gravitational / intermolecular forces*

allow 'electron(s) attracted less easily'

electron(s) gained less easily

"outer / last / final" must be mentioned once, otherwise max 2 marks.

accept converse for chlorine throughout where clearly stated

3

(e) (i) white precipitate **or** white solid

ignore names of chemicals

1

(ii) cream precipitate **or** cream solid

allow pale yellow / off-white precipitate / solid

ignore names of chemicals

1

[10]

M4. (a) not broken down by microorganisms **or** not bio-degradable

accept alternative answers such as:

do not rot / corrode / fade / react with atmosphere etc

any answers which imply the inertness or non-biodegradability of this plastic

accept they don't react, they are 'inert'

ignore rusting

*do **not** accept weathering*

1

(b) (i) (have a) double bond **or** do not have maximum number of (hydrogen) atoms attached

accept can add / react with hydrogen

accept can take part addition reactions

*do **not** accept it is a double bond*

*do **not** accept additional reactions*

*do **not** accept has 'spare' / 'free' bond*

*do **not** accept alkene alone*

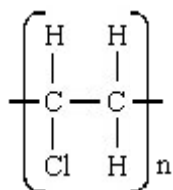
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(ii) single bond between carbon atoms

1

all atoms correct + 2 'linking' bonds

(linking bonds need not go through bracket)



1

n moved to bottom right of bracket i.e. is below $\frac{1}{2}$ way on the right

first 2 marks are possible for chain structures

accept $[-\text{CHCl}-\text{CH}_2-]_n$

1

(iii) many molecules **or** many monomers

1

joined / bonded / linked **or** form long
chain molecules / large molecules **or** to
form a long chain polymer

*accept many alkenes **or** many (ethene) molecules*

*do **not** accept many ethene alone etc.*

to form a long polymer is not enough for 2nd mark

1

(iv) no other substances formed

(A + B → C)

allow because double bond breaks so other atoms can add

allow one product only

*do **not** accept saturation occurs*

1

[8]

M5. (a) (i) by heating

pressure is neutral

using a catalyst/pot/ceramic/porcelain/aluminium oxide

1

(ii) use bromine water/(alkaline) permanganate

accept bromine

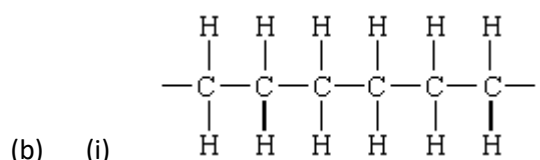
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alkene makes bromine go colourless or lose its colour

accept alkane does not change the red/orange colour of bromine

not change colour/goes clear

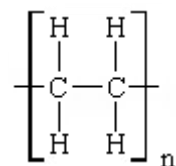
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either of these must show bonds at end

1

or



not H on ends

allow 3 instead of n not any other number

(ii) poly(ethene) – brackets not essential

accept polythene

1

(iii) **large amount** of waste polymer/poly(ethene)/polythene/litter
*accept large amount of crude oil **or** finite resource used*

1

it is not biodegradable

accept it does not

decompose/decay/break down

it causes pollution/it creates toxic

fumes when burnt are neutral

not *it is not recyclable*

2

[8]

M6. air/gases
oxygen O₂
chemical change/reaction
gases
carbon dioxide/CO₂
water [vapour] H₂O
(heat or heat energy or energy)
for 1 mark each

[7]

M7. (a) fuels
heat – allow light
for 1 mark each

2

(b) gases
for 1 mark

1

[3]